

Natural Abundance ¹⁷O DNP Two-Dimensional and Surface-Enhanced NMR Spectroscopy

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Supporting Information

ABSTRACT: Due to its extremely low natural abundance and quadrupolar nature, the ¹⁷O nuclide is very rarely used for spectroscopic investigation of solids by NMR without isotope enrichment. Additionally, the applicability of dynamic nuclear polarization (DNP), which leads to sensitivity enhancements of 2 orders of magnitude, to ¹⁷O is wrought with challenges due to the lack of spin diffusion and low polarization transfer efficiency from ¹H. Here, we demonstrate new DNP-based measurements that extend ¹⁷O solid-state NMR beyond its current capabilities. The use of the PRESTO technique instead of conventional ¹H-¹⁷O cross-polarization greatly improves the sensitivity and enables the facile measurement of undistorted line shapes and two-dimensional ¹H-¹⁷O HETCOR NMR spectra as well as accurate internuclear distance measurements at natural abundance. This was applied for distinguishing hydrogen-bonded and lone ¹⁷O sites on the surface of silica gel; the one-dimensional spectrum of which could not be used to extract such detail. Lastly, this greatly enhanced sensitivity has enabled, for the first time, the detection of surface hydroxyl sites on mesoporous silica at natural abundance, thereby extending the concept of DNP surface-enhanced NMR spectroscopy to the ¹⁷O nuclide.

The chemistry of oxygen is ubiquitous in all areas of the chemical sciences, from oxide-based materials, such as oxide catalysts and ceramics, to biochemistry, where hydrogen bonding to oxygen plays an important role in protein folding, for example. Nonetheless, while the solid-state nuclear magnetic resonance (SSNMR) spectroscopy of the other elements of importance in the chemical sciences, such as ¹H, ^{6/7}Li, ¹¹B, ¹³C, ¹⁵N, ²⁷Al, ²⁹Si, and ³¹P, to name a few, has flourished into a prominent characterization technique, ¹⁷O SSNMR investigations have failed to gain widespread use. The lagging development in ¹⁷O SSNMR spectroscopy is mainly due to the low natural isotopic abundance of ¹⁷O nuclei (NA = 0.038%), which renders the NMR signals unobservable unless tremendous experiment time is invested.¹ Additional challenge is posed by the quadrupolar nature of the ¹⁷O nuclide (I = 5/2), which results in line broadening that cannot be removed by magic angle spinning (MAS). Most ¹⁷O SSNMR studies have then opted for isotopic enrichment,² which is not always possible.

Recently, however, it was demonstrated that ¹⁷O SSNMR spectra could be acquired in a reasonable amount of time by using

dynamic nuclear polarization (DNP).³ In DNP, the nuclear magnetization is enhanced by transferring polarization from unpaired electrons to the nuclei of interest by the application of high-power microwaves near the electron Larmor frequency.⁴ This approach can yield signal-to-noise enhancements up to γ_{e}/γ_{n} , i.e., 658 for ¹H and 4855 for ¹⁷O, potentially surpassing the sensitivity that can even be obtained by ¹⁷O isotopic enrichment alone.

A particularly exciting application of DNP is for the selective enhancement of surface species that are often responsible for most of the chemistry of materials. This has been termed DNP surface-enhanced NMR spectroscopy (DNP-SENS)⁵ and has been applied to characterizing surface-supported catalysts and nanoparticles. The extension of DNP-SENS to include ¹⁷O NMR is highly desirable, for example, due to the importance of oxide supports in heterogeneous catalysis. However, further advances in sensitivity are necessary to enable the detection of such weak signals.

In the studies of small nanoparticles,³ or ¹⁷O-enriched molecular species in solutions, for which the ¹⁷O spins can be in proximity to the radicals,^{6 17}O nuclei have been directly polarized by DNP. Direct ¹⁷O DNP has also been performed in single crystals containing intrinsic paramagnetic impurities.⁷ Most recently, direct ¹⁷O DNP enhancements of as high as 115 have been obtained in biological solids using trityl radicals.⁶ However, direct ¹⁷O DNP of nonenriched samples has a major drawback in that spin diffusion within a material is too slow for the entire sample to be polarized.⁸ In samples containing hydrogen, it is advantageous to hyperpolarize ¹H, utilize ¹H-¹H spin diffusion, and subsequently transfer the ¹H magnetization to ¹⁷O nuclei within the bulk.^{3,9} The signal-to-noise enhancements that can be obtained using modern polarizing agents are also much larger when polarizing ¹H than 17 O; ¹⁰ note that an enhancement as low as 40 on ¹H translates to an enhancement of 295 for ¹⁷O, if we assume perfect polarization transfer.

The transfer of magnetization from ¹H to a quadrupolar nucleus, such as ¹⁷O, by cross-polarization (CP) presents, however, considerable challenges due to the unfavorable spin dynamics during spin locking and the CP process itself.¹¹ The poor spin locking characteristics of the central transition (m = 1/2 to -1/2 transition) of a quadrupolar nucleus under MAS conditions generate severe line shape distortions,¹² hampering the analysis of the spectra, and often also leads to sensitivity losses as opposed to gains.¹³ In order to take full advantage of the

Received:
 April 15, 2015

 Published:
 June 22, 2015



Figure 1. Pulse sequences to acquire natural abundance ¹⁷O NMR spectra with DNP: (a) CP-QCPMG, (b) PRESTO-QCPMG, and (c) PRESTO-QCPMG-HETCOR.

sensitivity improvements afforded by DNP, it is then necessary to utilize polarization transfer methods that overcome these two main issues associated with CP. The phase-shifted recoupling effects a smooth transfer of order (PRESTO) polarization transfer technique is such a method (see Figure 1 for the pulse sequence diagram).¹⁴ The PRESTO pulse sequence discards the problematic ¹⁷O spin locking pulse, and instead symmetry-based singlequantum heteronuclear recoupling is applied to ¹H spins, while a simple Hahn echo is performed at the ¹⁷O Larmor frequency. In our case, the $R18_1^7$ heteronuclear recoupling sequence was used,¹⁵ which also performs homonuclear decoupling, and the sensitivity is further enhanced by the application of quadrupolar Carr-Purcell-Meiboom-Gill (QCPMG) detection.¹⁶ Note, however, that alternative recoupling sequences may be used at higher MAS frequencies, such as R1825, to reduce the RF power requirements (Table S1). Additional details about the pulse sequences and the setup of all experiments are given in Supporting Information.

A comparison of the ¹⁷O CP-QCPMG and PRESTO-QCPMG SSNMR spectra of an ¹⁷O-enriched sample of silica gel are shown in Figure S1. It can be seen that, as is well documented, ¹² CP leads to a depletion of the signals at lower frequency due to the poor spin locking of those isochromats, whereas PRESTO yields the undistorted line shape and has an enhanced sensitivity, by a factor of 3.

We then acquired natural abundance ¹⁷O SSNMR spectra of $Mg(OH)_2$ and $Ca(OH)_2$ using both the CP-QCPMG and PRESTO-QCPMG techniques (Figure 2). The samples were impregnated with a 16 mM 1,1,2,2-tetrachloroethane solution of the state-of-the-art TEKPol biradical¹⁷ and spun to 12.5 kHz at a temperature of 105 K. Note that in these two samples, the ${}^{1}H-{}^{1}H$ spin diffusion propagated the magnetization into the bulk of the materials. As can be seen, the PRESTO technique provided signal enhancements on the order of 2-11, depending on the spikelet, and an overall sensitivity enhancement of 5, when compared to CP. This improvement means that DNP-enhanced ¹⁷O SSNMR experiments at natural abundance that would take a day to perform using CP can be performed in an hour using PRESTO. Additionally, both singularities of the powder pattern can be observed by PRESTO, something that cannot be done using the CP spectra, enabling the determination of the chemical shift and quadrupolar coupling parameters;¹⁸ this is more clearly seen in the spectra reconstructed from the weighed sum of the echoes.¹⁹ The intensities of the singularities can also be used to extract information regarding the relative orientation of the dipolar and EFG tensors.^{14b} SIMPSON²⁰ simulations of the PRESTO spectra using the known chemical shift and EFG tensor parameters²¹ are also shown in Figure 2 and are in good agreement with the experiment.

The PRESTO-QCPMG pulse sequence can also be modified to perform ${}^{1}\text{H}-{}^{17}\text{O}$ heteronuclear correlation (HETCOR) experiments (Figure 1c),²² where frequency-switched Lee– Goldburg homonuclear decoupling is applied during t₁ to enhance ${}^{1}\text{H}$ resolution.²³ We have indeed acquired the ${}^{1}\text{H}-{}^{17}\text{O}$ HETCOR spectra for Mg(OH)₂ and Ca(OH)₂, at natural abundance, using this approach. The spectra in Figure 3 show that the ${}^{17}\text{O}$ signal is mostly correlated to the hydroxyl resonance (located at 0 and 1 ppm for Mg(OH)₂ and Ca(OH)₂, respectively)²⁴ and that polarization is not transferred directly from the solvent (C₂H₂Cl₄). This is to be expected since PRESTO is mainly sensitive to short internuclear distances.¹⁴ ${}^{17}\text{O}-{}^{1}\text{H}$ correlation spectra have also been acquired using ${}^{1}\text{H}$ detection in



Figure 2. Natural-abundance, DNP-enhanced, ¹⁷O MAS NMR spectra of $Mg(OH)_2(a-c)$ and $Ca(OH)_2(d-f)$ acquired using the CP-QCPMG (c and f) and the PRESTO-QCPMG schemes (b and e). Both the spikelet (bottom) and reconstructed (top) spectra are shown. Signal to noise improvements from 2 to 11 (depending on the spikelet) are observed using PRESTO. SIMPSON simulations of the PRESTO MAS spectra are shown in (a) and (d). All spectra were acquired in 3.4 h.



Figure 3. Natural-abundance, DNP-enhanced, ${}^{1}H{-}^{17}O$ MAS PRESTO-QCPMG-HETCOR NMR spectra of Mg(OH)₂ (top) and Ca(OH)₂ (bottom) acquired in 4.5 h. Projections along the ${}^{1}H$ dimension show that polarization is transferred mainly from the hydroxyl proton and not the solvent.

enriched samples; 2d however, this would lead to t_1 noise at natural abundance, preventing the detection of 17 O.

By incrementing either of the two dipolar recoupling periods (τ_0 or τ_1 , see Figure 1b) in the PRESTO experiment, while keeping the other fixed, it is also possible to easily measure the ${}^{1}\text{H}{-}^{17}\text{O}$ dipolar coupling and thus determine the internuclear distance. 14 By decreasing the QCPMG period, so as to collapse all the signal intensity into a single spikelet, and incrementing the first recoupling block (τ_0), so as to maintain an echo delay of a single rotor period, PRESTO dipolar oscillations were measured in 30 min for Mg(OH)₂ and Ca(OH)₂ at natural abundance (Figure 4a,b). The resulting ${}^{1}\text{H}{-}{}^{17}\text{O}$ dipolar coupling constants (R_{DD}) of 18 ± 1 and 17 ± 1 kHz for Mg(OH)₂ and Ca(OH)₂,



Figure 4. DNP-enhanced ¹H–¹⁷O PRESTO-QCPMG dipolar oscillations from $Mg(OH)_2$, $Ca(OH)_2$, and ¹⁷O-enriched silica gel. SIMPSON simulations of the oscillations using the dipolar coupling strength indicated on the figure are shown in red. In (c), a sum of curves corresponding to lone and hydrogen-bonded silanol groups was inferred.

respectively, correspond to O–H bond lengths of 0.97 \pm 0.02 and 0.98 \pm 0.02 Å. These values are in remarkable agreement with the neutron diffraction structures of 0.96 and 0.94 Å for Mg(OH)₂ and Ca(OH)₂, respectively.²⁵ A similar approach has been used to study hydrogen bonding to oxygen in organic species.²⁶ In those cases elongated O–H bond lengths were measured due to the effects of vibrational averaging at room temperature.

When this experiment was performed on a ¹⁷O-enriched silica gel, the oscillations could not be simulated using only a single dipolar coupling constant (see Figure 4c). Instead, the curve corresponds to a sum of signals arising from hydrogen-bonded and lone surface silanol groups, having effective dipolar coupling constants of 17 ± 1 and 8.5 ± 1 kHz. The 50% reduction in dipolar coupling strength for some sites likely originates from partial averaging of the dipolar coupling under librational motions;²⁷ similar effects have been seen using deuterium NMR of surface silanol groups in silica gel at these temperatures.²⁸ In fact, such a model would indeed yield the fitted curve in Figure 4c. This type of motion is not structurally permitted in Mg(OH)₂ and $Ca(OH)_2$. The signals arising from the strongly dipolar coupled pairs must then represent rigid silanols, whose librations are restricted by the frozen solvent or by weak hydrogen bonding to neighboring silanol groups or to water molecules at the surface. In principle, the reduction in dipolar coupling could also originate from a large elongation of the O–H bond to 1.2 Å from a very strong hydrogen bond; this is, however, unlikely due to the prevalence of the site and the rarity of such interactions. The ability to study ¹H-¹⁷O dipolar oscillations is very significant because this technique affords a straightforward way of studying the interactions between the surface of a silica support and guest molecules, even for multiple chemical environments with similar chemical shift and EFG parameters.^{2d}

Due to the remarkable sensitivity afforded by DNP-enhanced PRESTO-QCPMG NMR, we were able to perform the first natural abundance ¹⁷O DNP-SENS experiment.⁵ The ¹⁷O DNP-SENS NMR spectrum of SBA-15, a mesoporous silica sample having a surface area of 600 m²/g and a pore diameter of 9.6 Å,²⁹ acquired in 14 h is shown in Figure 5. This spectrum demonstrates



Figure 5. Natural-abundance DNP-SENS $^{17}\mathrm{O}$ MAS NMR spectrum of the mesoporous silica SBA-15, acquired using the PRESTO-QCPMG pulse sequence in 14 h.

that by utilizing state-of-the-art equipment, biradicals, and pulse sequences, it will be possible to measure $^1\mathrm{H}-^{17}\mathrm{O}$ internuclear distances of catalyst supports, thereby enabling the characterization of hydrogen bonding and dynamics on the surface of catalytically important materials at natural abundance.

In summary, the scope of natural abundance DNP-enhanced ¹⁷O SSNMR is greatly enhanced via the application of improved

polarization transfer. The improved sensitivity of PRESTO decreases experiment times by over an order of magnitude, when compared to CP, such that ¹⁷O DNP NMR is now sufficiently sensitive for the application of ¹⁷O DNP-SENS at natural isotopic abundance. Additionally, reliable line shapes can be obtained from the spectra, and the experiment can be trivially altered to measure HETCOR spectra and ¹H $^{-17}$ O internuclear distances. This has enabled the detection of hydrogen-bonded and non-hydrogen-bonded silanol groups, whose dynamic motions can be probed by this approach, at the surface of silica gel. Such distinction cannot be made otherwise. These developments may facilitate the widespread application of ¹⁷O DNP-SENS NMR, as well as other techniques relying on transfer of polarization between spin-1/2 and quadrupolar nuclei, for probing the structure and dynamics in a variety of materials.

ASSOCIATED CONTENT

Supporting Information

The PRESTO-QCPMG pulse sequence is described. Details regarding the experiment setup, the simulations, and the sample preparations are given. The spectra acquired for an ¹⁷O-enriched silica gel are shown. The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/ jacs.5b03905.

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Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We thank M. Carravetta and I. M. Haies for sharing their PRESTO-II pulse program, I. I. Slowing for synthesizing silica SBA-15, and J. Stebbins for ¹⁷O-enriched silica gel sample. This research is supported by the U.S. Department of Energy (DOE), Office of Science, Basic Energy Sciences, Division of Materials Science and Engineering (methods development), and Division of Chemical Sciences, Geosciences, and Biosciences (specific applications). Support for F.P. is through a Spedding Fellowship funded by the LDRD program. Ames Laboratory is operated for the DOE by Iowa State University under Contract No. DE-AC02-07CH11358.

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